## PHY 3107, Spring 2017, Homework #6 due Tuesday, March 7

- 1.) Two spin-zero bosons are in a one-dimensional infinite well. [The wavefunction for a single particle in an infinite well is  $\psi(x) = \sqrt{2/L} \sin(n\pi x/L)$ .] a) Write the wavefunction for the system consisting of these two particles. b) What is the lowest (or ground-state) energy of the system?
- 2.) Two electrons each with spin- $\frac{1}{2}$  are in a one-dimensional infinite well. [The wavefunction for a single particle in an infinite well is  $\psi(x) = \sqrt{2/L} \sin(n\pi x/L)$ .] a) Write the wavefunction for the system consisting of these two particles. b) What is the lowest (or ground-state) energy of the system? [Hint: Ignore spin and construct an anti-symmetric spatial wavefunction for the first part. But for the second part, remember that the two electrons cannot be in the same spatial wavefunction!]
- **3.)** The outer electron of the potassium atom occupies the  $4s^{1}$  atomic level. You measure a value for the ionization energy of this electron of 4.35 eV. Using this information and example 9.6 in your text, find the value of  $Z_{eff}$  for the  $4s^{1}$  electron in potassium.
- **4.)** Let's look at oxygen. A) write out the electronic configuration for the atom oxygen (charge 8). B) Does oxygen have the outermost shell filled? C) For each of the 8 electrons, write the quantum numbers n, l,  $m_l$ ,  $m_s$ , j,  $m_j$ . [Note: The final 2 are superfluous.]