# PHYS 4134, Fall 2016, Homework \#6 

## Due Wednesday, October 192016

## 1. Scintillators

The Thallium-doped Sodium Iodide scintillator, $\mathrm{NaI}(\mathrm{Tl})$, is example of an inorganic scintillator. Calculate the radiation energy in eV deposited in the scintillator per photoelectron. [See problem 6.6]
A) The average photon from the de-excitation of an electron from the conduction band to the valence band in the crystal is 2.9 eV ; calculate the photon energy in eV . [Hint: it should be 2.9 eV.$]$
B) $10 \%$ of the electron de-excitations result in the emission of a fluorescent photon, and $90 \%$ of these photons reach the photocathode of a PMT which has an efficiency of 0.15 photoelectrons per incident photon. Taking a single photon coming from an atomic deexcitation, calculate the radiation energy E (in eV ) deposited in the scintillator per photoelectron. [See also problem 6.6]
C) You are using a Hamamatsu R594 PMT (you can see the characteristics, including the number of dynodes, at the product web page http://www.hamamatsu.com/us/en/R594.html ). If the gain is $X^{N}$ where N is the number of dynodes, calculate $X$, the secondary electron emission coefficient if the PMT is operated at 1500 V where the gain is $7.4 \times 10^{5}$. [This "secondary electron emission coefficient" represents the average number of electrons knocked out of every stage by an incident electron.]
D) If the capacity C of the anode is 50 pF , calculate the output voltage pulse if a photon of energy $\mathrm{E}=200 \mathrm{keV}$ is stopped in the scintillator crystal. [Again, see problem 6.6]

## 2. Neutron Counters

Plastic scintillators are examples of inorganic scintillators. In this case, we are looking at Bicron BC-400, a plastic consisting of carbon and hydrogen and more commonly known as polyvinyltoluene, or PVT. Although technically in the form $\left(\mathrm{C}_{27} \mathrm{H}_{30}\right)_{\text {n }}$, we will model it as the simpler $(\mathrm{CH})_{\mathrm{n}}$. Take the neutron scattering cross section for carbon and hydrogen to be equal and 2 barns in magnitude. The product sheet (available under the specifications tab at the website http://www.crystals.saint-gobain.com/Plastic_Scintillators.aspx provides the density. We will use a 5 cm thick piece. Your detector will only respond to scattering from protons, meaning you first have to scatter the neutrons from the C or H , and then detect the scattering protons. You can take the absorption cross section to be negligible compared to the scattering cross section. [See also problem 6.7]

## 3. Fission fragmentation

A nucleus of atomic number $A$ decays by fission into two unequal fragments, $f_{1}$ and $f_{2}$. These fragments are detected in two detectors, each placed an equal distance on opposite sides of the fission source. If the flight time for $f_{l}$ is $30 \%$ less than that for $f_{2}$, find: a) the ratio of the masses, and $b$ ) the ratio of the kinetic energies of the two fragments. [See also problem 6.8]

