

Bose-Einstein Condensation

- Integer spin particles with large mass
- Particles make the Bose-Einstein gas
- Example of such particle is ^4He atoms
- Quantum Effects expected at very low $T \rightarrow$ Bose-Einstein Condensation
- BEC - related to superfluidity - Liquid Helium
- No Exclusion Principle
- Consider Ideal Boson gas consisting of N -particles in the volume V at T

$$f(\epsilon) = \frac{N(\epsilon)}{g(\epsilon)} = \frac{1}{e^{\frac{\epsilon - \mu}{kT}} - 1} \quad (1)$$

i) μ as it depends on T

- Adopt convention of Ground State Energy to be 0
- At $T=0$ all Bosons N are in the Ground state
- Setting $\epsilon \rightarrow 0$ $f(\epsilon)$ makes sense if μ is negative
- at $T=0$ $\mu=0$ and $\mu < 0$ $T > 0$
- At high temperatures - in the Classical Limit \rightarrow Maxwell Boltzmann distribution

$$f(\epsilon) = e^{-\frac{\epsilon - \mu}{kT}} \quad \text{check this}$$

$$Z = \sum_j g_j e^{-\epsilon_j/kT} = \int_0^\infty g(\epsilon) e^{-\epsilon/kT} d\epsilon$$

$$e^{-\frac{\epsilon}{kT}} \cdot e^{\frac{\mu}{kT}} = f(\epsilon) = \frac{N(\epsilon)}{g(\epsilon)}$$

$$g(\epsilon) e^{-\frac{\epsilon}{kT}} \cdot e^{\frac{\mu}{kT}} = N(\epsilon)$$

$$\left(g(\epsilon) e^{-\frac{\epsilon}{kT}} d\epsilon \right) \cdot e^{\frac{\mu}{kT}} = N(\epsilon) d\epsilon = N$$

$$\frac{\mu}{kT} = \frac{N}{Z}$$

$$\frac{\mu}{kT} = \ln \frac{N}{Z}, \quad \mu = -kT \ln \frac{Z}{N}$$

Phase space of non-relativistic Particle

$$dn = \frac{V d^3p}{(2\pi\hbar)^3} = \frac{V p^2 dp d\Omega}{(2\pi\hbar)^3} \quad p = \sqrt{2m\varepsilon}$$

$$= \frac{V 2m\varepsilon \sqrt{2m} \cdot \frac{1}{2} d\varepsilon d\Omega}{(2\pi\hbar)^3} = \frac{4\pi V \sqrt{2m} \cdot m^{3/2} d\varepsilon}{h^3}$$

$$g(\varepsilon) dn = \delta_3 \cdot V \cdot \frac{4\sqrt{2}\pi}{h^3} m^{3/2} \varepsilon^{1/2} d\varepsilon$$

Calculate Z $\delta_3 = 1$

$$Z = \frac{V \cdot 4\sqrt{2}\pi m^{3/2}}{h^3} \int_0^\infty \varepsilon^{1/2} e^{-\frac{\varepsilon}{kT}} d\varepsilon \quad \text{HW}$$

$$Z = \left(\frac{2\pi m kT}{h^2} \right)^{3/2} V$$

$$\frac{\mu}{kT} = -\ln \left[\left(\frac{2\pi m kT}{h^2} \right)^{3/2} \frac{V}{N} \right] \quad (2)$$

One kilomole of Boron gas of ^{10}B at $T=273$

$$\mu_H = \frac{6.65 \times 10^{-27} \cdot (1.38 \times 10^{-23}) \cdot 273}{(6.63 \times 10^{-34})^2} \left(\frac{22.4}{6.02 \times 10^{23}} \right)$$

$$\mu = -0.23 \text{ eV}$$

Comparing with average energy of monoatomic gas $\varepsilon = \left(\frac{3}{2}\right)kT = 0.035 \text{ eV} \quad 273 \text{ K}$

$$\text{So } \frac{(\varepsilon - \mu)}{kT} = 1.5 + 0.4 = 1.9$$

$$\text{Substituting in } f(\varepsilon) = e^{-\frac{(\varepsilon - \mu)}{kT}} = \frac{9.2 \times 10^{-7}}{\text{Total Boron}}$$

where $\mu \rightarrow$
Assumption
is valid

from (25)

$$e^{-\frac{\mu}{kT}} = \left(\frac{2\pi m k T}{h^2} \right)^{3/2} \frac{V}{N}$$

\Rightarrow increases with T

\Rightarrow decreases with $N = \frac{1}{V}$

Distribution of N with μ - function of N/V Temperature

\Rightarrow continuum approximation

$$\int_0^\infty n(\epsilon) d\epsilon = \int_0^\infty f(\epsilon) g(\epsilon) d\epsilon = N$$

$$g(\epsilon) d\epsilon = \frac{4\sqrt{2}\pi V}{h^3} m^{3/2} \epsilon^{1/2} d\epsilon$$

$$N = 2\pi V \left(\frac{2m}{h^2} \right)^{3/2} \int_0^\infty \frac{\epsilon^{1/2} d\epsilon}{e^{\frac{\epsilon - \mu}{kT}} - 1} \rightarrow \text{Numerically}$$

\Rightarrow problem that this does not account the $\epsilon=0$ contribution

Introduce $N = N_0 + N_{ex}$ this is N_{ex}

at $T \rightarrow 0$ $N_0 \approx N$

we can put $\epsilon=0$ in $E_F(0)$

$$f(\epsilon) \stackrel{T \rightarrow 0}{=} \frac{1}{e^{\frac{\epsilon}{kT}} - 1} \approx N \quad \text{since all particles are at } \epsilon=0$$

$$-\frac{\mu}{kT} = \ln(1 + \frac{1}{N}) \approx \frac{1}{N}$$

→ so $\frac{\mu}{kT} \sim \frac{1}{N} \rightarrow 0$ small

Thus in (3) we put $\frac{\mu}{kT} = 0$ $T \rightarrow 0$
 $x = \frac{\epsilon}{kT}$

$$N_{ex} = V \frac{2}{\sqrt{\pi}} \left(\frac{2\pi m kT}{h^2} \right)^{3/2} \int \frac{x^{1/2} dx}{e^x - 1}$$

$\xrightarrow{2.612 \frac{\sqrt{\pi}}{2}}$

$$N_{ex} = 2.612 \cdot V \left(\frac{2\pi m kT}{h^2} \right)^{3/2} \quad (4)$$

→ Bose Temperature T_B $N_{ex} = N$

$$N = 2.612 V \left(\frac{2\pi m k T_B}{h^2} \right)^{3/2} \quad (5)$$

$$T_B = \frac{h^2}{2\pi m k} \left(\frac{N}{2.612 V} \right)^{2/3}$$

→ $T > T_B$ all bosons are excited

→ $T < T_B$ Bosons → $E=0$ state
 until all $N = N_0$

$$\frac{N_0}{N} = 1 - \frac{N_{ex}}{N} \quad \text{from 4 and 5}$$

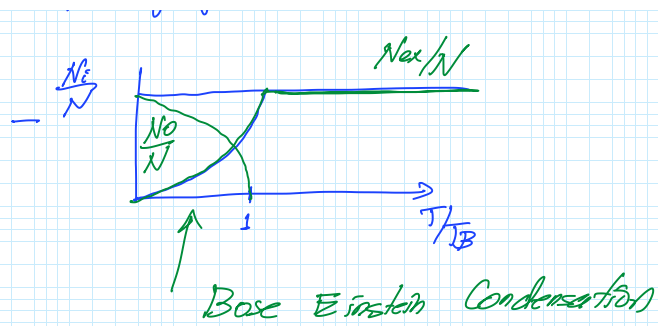
$$\frac{N_0}{N} = 1 - \left(\frac{T}{T_B} \right)^{3/2}$$

kilomole

→ Boson Gas ${}^4\text{He}$ 6.02×10^{23} ${}^4\text{He}$
 Volume $22.4 \times 10^{-3} \text{ m}^3$
 $m_{He} = 6.65 \times 10^{-27} \text{ kg}$

$$T_B = 0.036 \text{ K} \quad \text{HVD}$$

→ ${}^4\text{He}$ liquidifies $T_L = 4.2 \text{ K}$ 1 atm



— Found in 1955

\Rightarrow Properties of Boson Gas

— In the ground state do not contribute to the internal energy and nor to Heat Capacity

$$E = N_0 \epsilon_0 + N_{\max} \epsilon_{\max}$$

for $T < T_B$, N_0 large but $\epsilon_0 = 0$

for $T > T_B$, $N_0 = 0$

— For $T > T_B$ all N -s are in excited state

$$\text{Expect } U \rightarrow \frac{3}{2} N K T$$

$$\text{— For } T < T_B \quad N_{\max} = N \left(\frac{T}{T_B} \right)^{3/2}$$

— Assume each boson has thermal energy of the order of $K T$

$$U \approx N K T \left(\frac{T}{T_B} \right)^{3/2} \quad (T < T_B)$$

$$U \sim T^{5/2}$$

— Nearly exact result obtained

$$U = \int \epsilon N(\epsilon) d\epsilon$$

using $f(\epsilon) = \frac{N(\epsilon)}{g(\epsilon)} = \frac{1}{e^{\frac{\epsilon - \epsilon_0}{kT}} - 1}$ and

$$g(\epsilon) d\epsilon = \frac{4\pi V}{h^3} m^{3/2} \epsilon^{1/2} d\epsilon$$

$$U = 2\pi V \left(\frac{2m}{h^2} \right)^{3/2} \int_0^\infty \frac{\epsilon^{3/2} d\epsilon}{e^{\frac{\epsilon - \epsilon_0}{kT}} - 1}$$

— If we choose $\epsilon_0 \approx 0$ then $U \approx 0$
for $T < T_B$

— Setting $U=0$, substitution $x = \frac{\epsilon}{kT}$

$$U = \frac{2}{\sqrt{\pi}} kT \left(\frac{2\pi m kT}{h^2} \right)^{3/2} \int_0^\infty \frac{x^{3/2} dx}{e^x - 1}$$

$$\Gamma\left(\frac{5}{2}\right) = \frac{3\sqrt{\pi}}{4}$$

Riemann Zeta ζ

$$\zeta\left(\frac{5}{2}\right) = 1.34$$

— Using $T_B = \frac{h^2}{2\pi m k} \left(\frac{N}{2 \cdot \pi V} \right)^{2/3}$

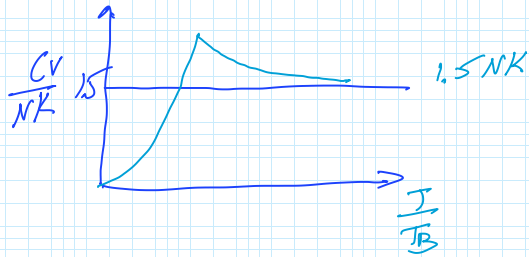
$$U = 0.77 N kT \left(\frac{T}{T_B} \right)^{3/2} (T < T_B)$$

— Heat Capacity:

$$C_V = \frac{dU}{dT} = 1.17 N k \left(\frac{T}{T_B} \right)^{3/2} (T < T_B)$$

$$C_V \rightarrow \frac{U_V}{dT} = 1.50 NK(T_B) \quad U \rightarrow$$

$$(V \sim T^{3/2})$$



— Entropy $T < T_B$

$$S = \int_0^T \frac{C_V dT'}{T'} = 1.28 NK \left(\frac{T}{T_B} \right)^{3/2} \quad T < T_B$$

$$S \rightarrow 0 \quad T \rightarrow 0$$

— F — Helmholtz Function

$$F = U - TS$$

$$F = -0.51 NK T \left(\frac{T}{T_B} \right)^{3/2} \quad (T < T_B)$$

— Pressure P

$$P = - \left(\frac{\partial F}{\partial V} \right)_{T,N}$$

$$T_B = \frac{h^2}{2\pi m K} \left(\frac{N}{2.602 V} \right)^{2/3}$$

$$F = -1.33 K T \left(\frac{2\pi m K T}{h^2} \right)^{3/2} \cdot V$$

$$P = 1.33 K T \left(\frac{2\pi m K T}{h^2} \right)^{3/2} \quad (T < T_B)$$

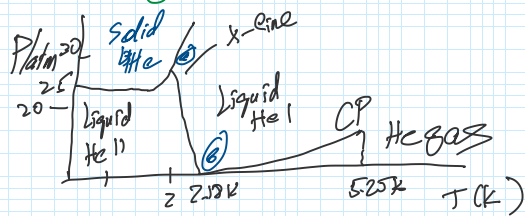
$$P = \frac{5}{6} n k T \quad T > T_B$$

So $P \propto T^2$ $\propto \frac{1}{V}$ $T \propto V$
independent of the volume

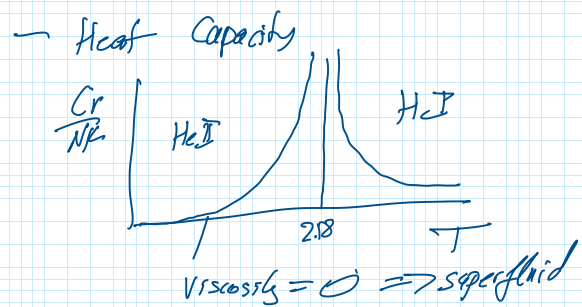
- This is because at $T=0$ $E=0$
have no momentum and therefore
make no contribution to the
pressure

⇒ Application to Liquid Helium

Phase Diagram of ^4He



- Critical Point of 5.25 K
- Unique Behavior at $T = 2.18 \text{ K}$
- Compressed isothermally $T \gg 2.18 \text{ K}$
it condenses to a liquid phase \downarrow
- Compressed at $T < 2.18 \text{ K}$
Liquid phase ^4He I
- Helium II - is superfluid
- X-line I and II Coexist
- He has 2 triple point (1) and (2)



He II ideal Boson Gas

λ - Bose T_B

He II \rightarrow No atoms

Kiloware $V = 27 \times 10^{-3} \text{ m}^3$

$$\frac{N}{V} = \frac{6.02 \times 10^{26}}{27 \times 10^{-3}} = 2.2 \times 10^{28} \text{ m}^{-3}$$

$${}^4\text{He mass} = 663 \times 10^{-27} \text{ kg}$$

$$T_B = 3.1 \text{ K}$$

\rightarrow compared to 2.18 K point

