



Transportation

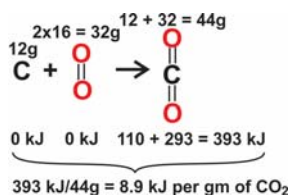
- Automobiles and airplanes are 20th Century icons of modernity
- Just as railroads and steamships were in the 19th Century
- And computers and the Internet are in the 21st Century
- The normal time from a marketable prototype to mature technology is 25-50 years
 - e.g., Wright flyer (1904) to DC3 (1936), or Boeing 707 (1958)
 - First locomotive (Salamanca, 1812) to 10⁴ km of track in UK (1850) or US Transcontinental railroad (1869)
 - Benz "Motorwagen (1886)" to Model T (1908)
- Even though automobiles are so 20th Century, everybody worldwide still wants one
- They and airplanes are nearly all powered by fossil fuel
- And emit lots of Green-House Gasses

Units of Measurement

- Use SI units
 - Kilogram = 2.2 pounds (mass)
 - Meter = 39.4 in (length; 1 foot ~ 30 cm = 0.3 m)
 - Second (time)
- Force = mass x acceleration = kg-m/s², measured in Newtons (Nt)
 - Gravity = 9.8 m s⁻² (meters per second squared)
- Energy is force acting through a distance = Joule = kg-m²s⁻²
- Power = energy/time = Joule/second = Watt = kg-m²s⁻³
- 1 Horsepower = 745.7 W
- In chemical reactions, mass is measured in moles
 - Avogadro's number = 6.22 x 10²³ atoms
 - Or the atomic weight in grams
 - C = 12 gm CO₂ = 12 + 2x16 = 44 gm, or H₂O = 2 + 16 = 18 gm
 - Or 22.4 liters (22.4 x 0.001 m³) of any gas
- We use prefixes kilo (k = 1,000) and mega (m = 1,000,000)

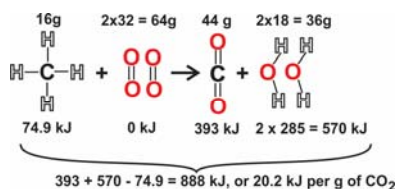
Burning Fuel

- Most transport fuels are coal or petroleum burned in air.
- For example **coal**, which is mostly carbon
- Burn one mole of coal with one mole of oxygen to make one mole of carbon dioxide



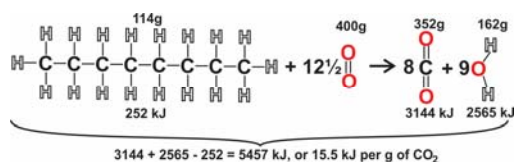
- So burning 12 gm of coal in a minute (60s) would release heat = 393000/(60x745.7) = 8.8 HP.
- Since steam engines are ~35% efficient, we would get only 3.1 HP
- Or 2300 W

How About Natural Gas?



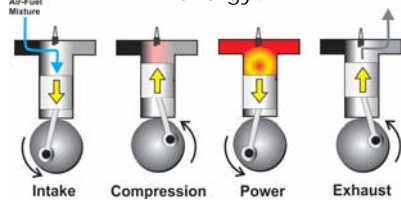
- A mole of methane weighs 16 gm, or 4 gm (¼) more than one of carbon
- But it produces more than twice as much energy
 - 888 kJ instead of 393 kJ
- Because of the heat released by making 2 moles of H₂O
- And it releases more than twice as much energy per gram of CO₂ emitted (20 vs 8.9 kJ/g)

How About Gasoline (Octane)?



- A mole of octane weighs ~7 times as much as one of methane
- And produces about 6 times as much energy
- Because there are closer to two hydrogen atoms per carbon atom than four
- And it releases 15 kJ of heat per g CO₂ (20 for methane and 8.9 for pure carbon)

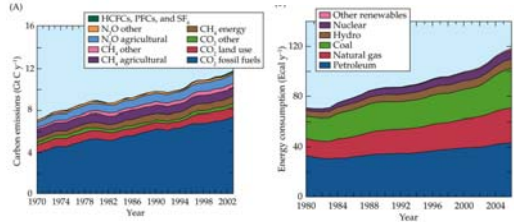
How do we convert chemical energy to mechanical energy?



- Four-stroke, "Air-Standard Otto Cycle"
 - Intake: Draws fuel and air into cylinder
 - Compression: Heats mixture by raising pressure
 - Power: Combustion of fuel increase pressure pushing piston down
 - Exhaust: Forces burned fuel out of the cylinder
- Automobile engines convert 25-30% of the energy in fuel into useful work
- Higher compression ratios increase efficiency

GHG Emissions and Energy Sources

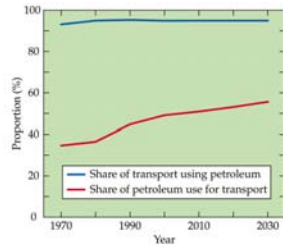
Green-House Gas Sources Energy Consumption (10¹⁸ cal/yr)



NOTE: 1 cal = 4.186 J or 1.163 x 10⁻³ W-hr

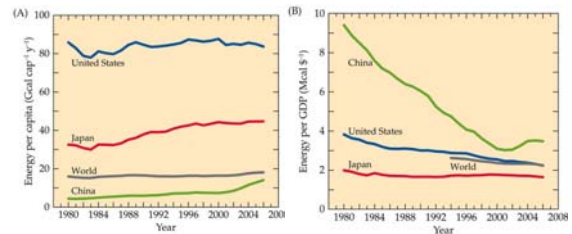
Petroleum and Transport

- More than 90% of 21st Century transport is fueled with petroleum
- The fraction of petroleum production used for transport will increase from 34% in 1970 to 57% by 2030.



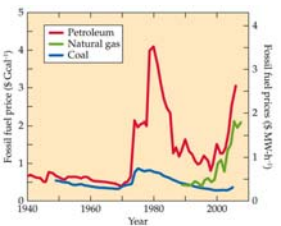
Representative Energy Usage by Population and GDP

Per capita energy usage: (Gcal yr⁻¹ = 132.2 W; 80 Gcal yr⁻¹ cap⁻¹ = 10 kW cap⁻¹)
 Energy per dollar of GDP: Mcal = 4.186x10⁶ J ~ 1.1 kW h.

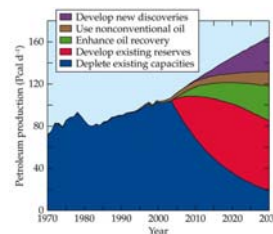


Prices and Sources of Fuels

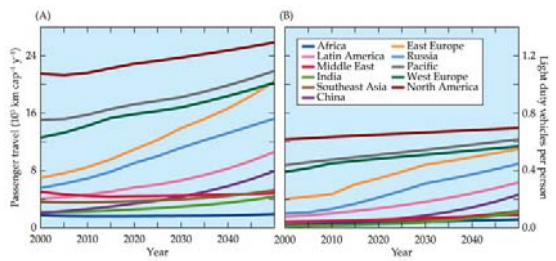
Prices of Energy from Petrol, Natural Gas and Coal (\$/Gcal)

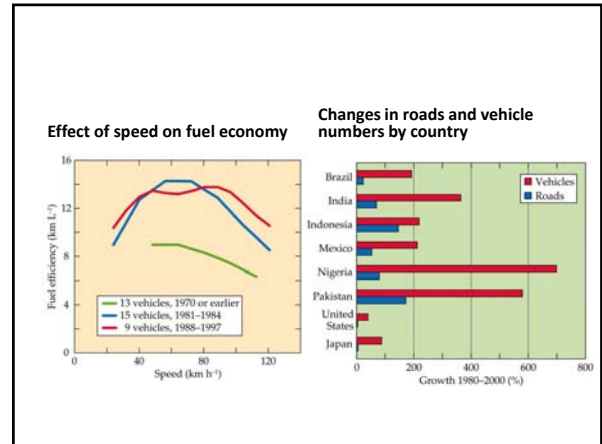
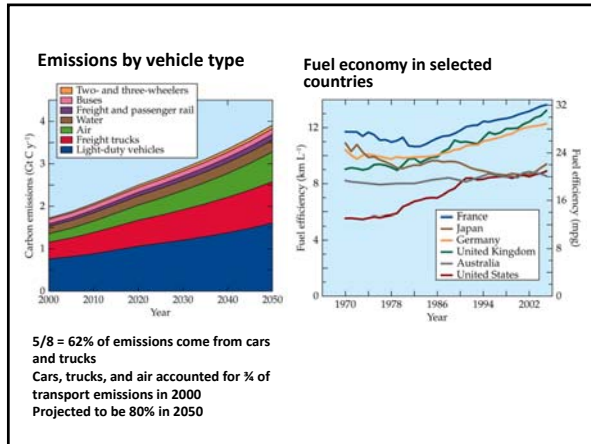


Sources of Petroleum



Travel and Vehicle Ownership by Region





How does it all fit together?

- People want heavier, more powerful cars
- In the developing world, people want more cars
- Fuel economy is becoming somewhat better because of better engineering
- Emissions from all modes of transportation (except busses) are increasing 2% per year
- Individuals have direct control over emissions from their cars
- Factors affecting emissions besides weight, power and numbers
 - Stop & go
 - Speed
 - Taxes
 - Gridlock, accidents & weather
 - Innovative power plants

The BIG PICTURE

- Worldwide, Road transportation accounts for ~10% of CO₂ emissions
- Sea and air transportation accounts for 2-3%
- But individuals have direct control (also in residences)
- Very efficient alternatives
 - Urban light rail
 - Freight rail
 - Containerships
 - Ocean liners???
- Note that **electrical generation** accounts for ~30% of emissions

Alternative Technologies

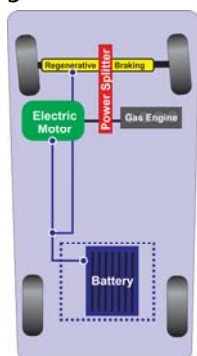
- Hydrogen Fuel Cell
- Gas Hybrid
- Plug-In Hybrid
- Biofuel
- Ethanol
 - Fermentation
 - Cellulosic
 - Butanol

Hydrogen Fuel Cell

- Reacts H₂ with O₂ in the presence of a catalyst to produce electric current and H₂O
- Hydrogen storage has low density
 - Liquid H₂
 - Metal hydrides
- Production of H₂
 - Electrolysis
 - From fossil fuel: CH₄ + H₂O → CO + 3H₂
- Downsides
 - Expense
 - GHG emissions from fuel production
 - Low energy density
 - Fire and explosion hazard
 - New distribution system required

- Have gas and electric motors
- Electric motor is also a generator
 - Powered by gas engine
 - And regenerative braking
- Gas engine is turned off except when car is moving
- Electric motor provides start up
- Gas engine runs for normal cruising
- Electric motor provides (lots of) extra torque for starting, hills, passing...
- Saves gas by not idling and by regenerative braking
- Plug-in has bigger battery for commuting and runs gas engine only when battery is low or for longer trips
- Can double fuel efficiency

Hybrid Cars



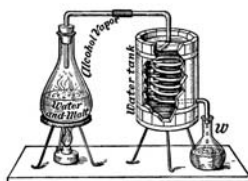
Biofuels: Ethanol From Maize

- Capture sunlight, CO_2 & H_2O to make Glucose stored in the grain.
- Ferment the glucose to make ethanol ($\text{CH}_3\text{CH}_2\text{OH}$) and some CO_2
- Distill the result to make fuel
- Burn the fuel to make energy and more CO_2
- Carbon neutral apart from energy inputs to growing processing and distribution



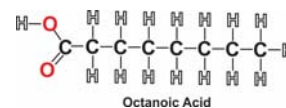
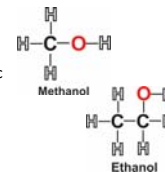
- In the US, Maize yields 10^4 kg per hectare
- Which can produce 0.37 L of $\text{CH}_3\text{CH}_2\text{OH}$ per kg
- Or 3700 L of ethanol per hectare
- US consumes 10^{12} liters of crude oil per year
- Assuming one-to-one replacement of petroleum with ethanol, that's 270,000 hectares
- Compared with 38,000 currently in maize production
- And 916,000 hectares of US land area

Critique



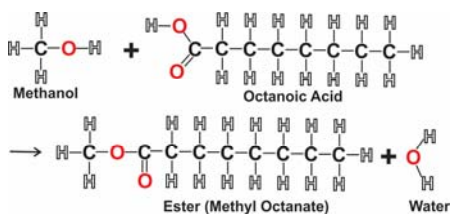
Some Organic Chemistry

- Some fuels are organic bases ($\text{pH} > 7$), called alcohols:
- In addition to alcohols, there are "fatty" organic acids ($\text{pH} < 7$) made up of a long hydrocarbon chain with a carboxyl group (COOH) at the end.
- Vinegar, CH_3COOH , is an example, even though it's not all that fatty
- Octanoic acid, which is related to Octane, is another.
 - Note the COOH on the left end



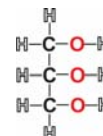
Esters

- Alcohols and fatty acids combine to form Esters.
 - Often have pleasant smells and are used in flavoring
 - But are a problem for biofuels



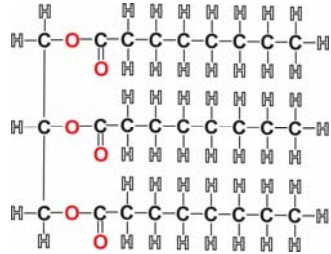
Glycerol

- Is a 3-carbon chain with a hydroxide on each carbon.
 - Which is to say an alcohol
 - Used in cosmetics and some foods
 - A triple-threat: Alcohol that can form an ester with three fatty acids at the same time.



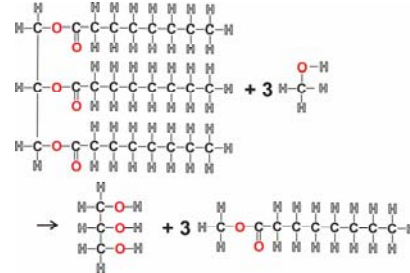
Triglycerides

- Glycerol forms a backbone with three fatty acids attached.
- Most biologically produced fats are triglycerides
- Because they have such a large molecular weight, they do not work in internal combustion engines



Transesterification

- Breaks up triglycerides into lower molecular weight oils that can be used as **Biodiesel**

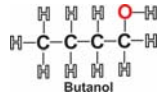


Can be used to reclaim the energy content of waste fats (i.e., cooking oil) but purpose grown oils are much less promising

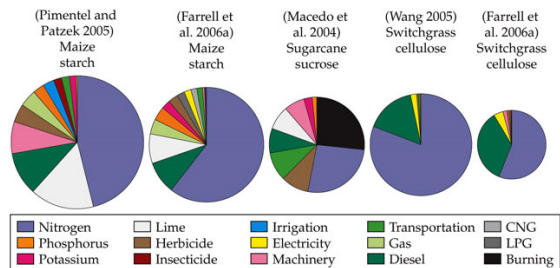
Switchgrass



- Switchgrass uses C4 photosynthesis
 - Tolerates drought and heat well
- Yields ~ 15,000 kg of dry biomass per hectare
 - Cellulose and lignin
- About half the energy input of Maize per hectare
- Clostridium acetobutylicum can ferment cellulose into butanol
- Process also works for waste paper, garden trash, ...
- Butanol can go right into your gas tank

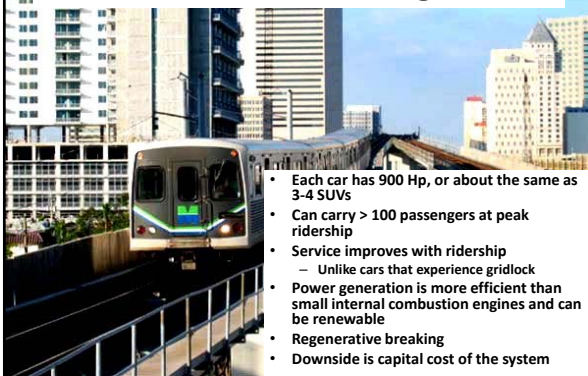


Comparison of Energy Inputs for Biofuels



Gram carbon equivalents per L of ethanol: 0.31,0.21,0.10,0.11,0.05

In Praise of Urban Light Rail



- Each car has 900 Hp, or about the same as 3-4 SUVs
- Can carry > 100 passengers at peak ridership
- Service improves with ridership
 - Unlike cars that experience gridlock
- Power generation is more efficient than small internal combustion engines and can be renewable
- Regenerative braking
- Downside is capital cost of the system

SUMMARY

- Fossil fuels account for 69% of GHG emissions
- Transportation contributes 21% globally (US:33%)
- Petroleum is principal fuel
- Methane produces 50% as much CO₂ per Joule as coal. Octane, etc., about 75%
- Cars are becoming more efficient, but...
- Everybody wants more, bigger & more powerful cars
- Trains & ships can be very fuel efficient
- But not airplanes
- Hybrids can double fuel efficiency
 - No idling and regenerative braking
- Biofuels
 - Rich peoples fuel competing with poor people's food
 - Ethanol from grain
 - Biodiesel from (waste) fats
 - Switch grass and Clostridium acetobutylicum